

Asphaltene: A General Introduction

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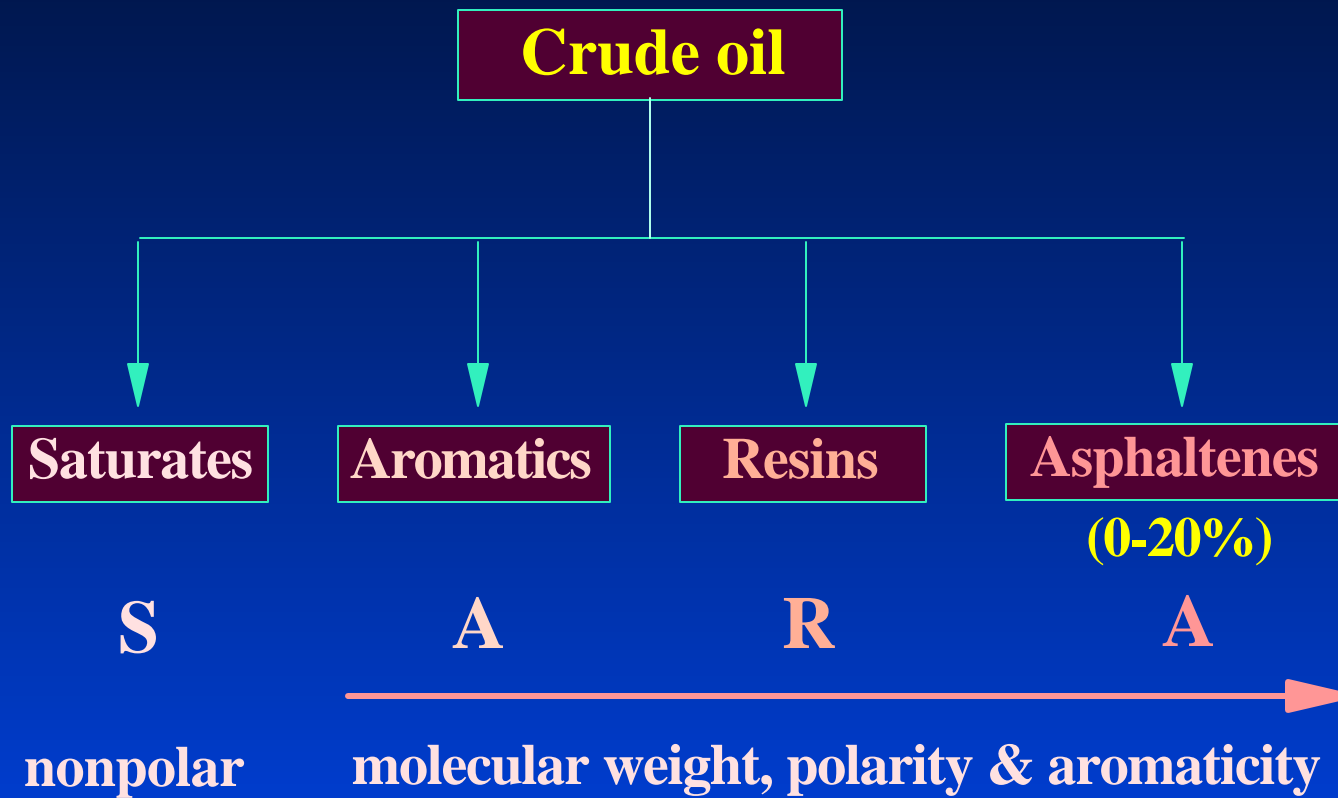
**P & Sc Group
PRRC, New Mexico Tech**

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Outline

- 👉 **What is asphaltene?**
- 👉 **Why to study asphaltene?**
- 👉 **How to characterize asphaltene stability in crude oil?**

Asphaltenes are the most heavy and polar components in crude oil

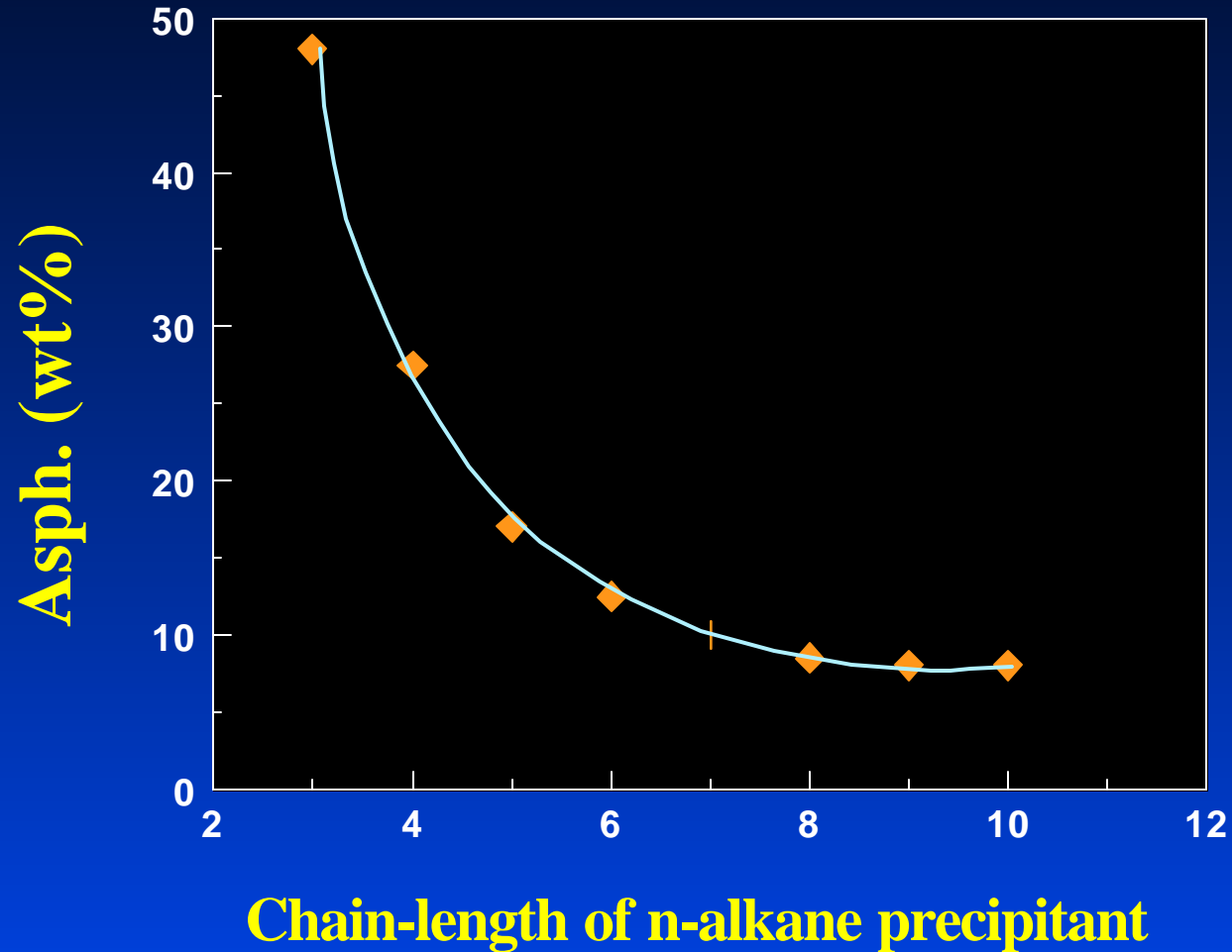


Separating asphaltene from crude oil

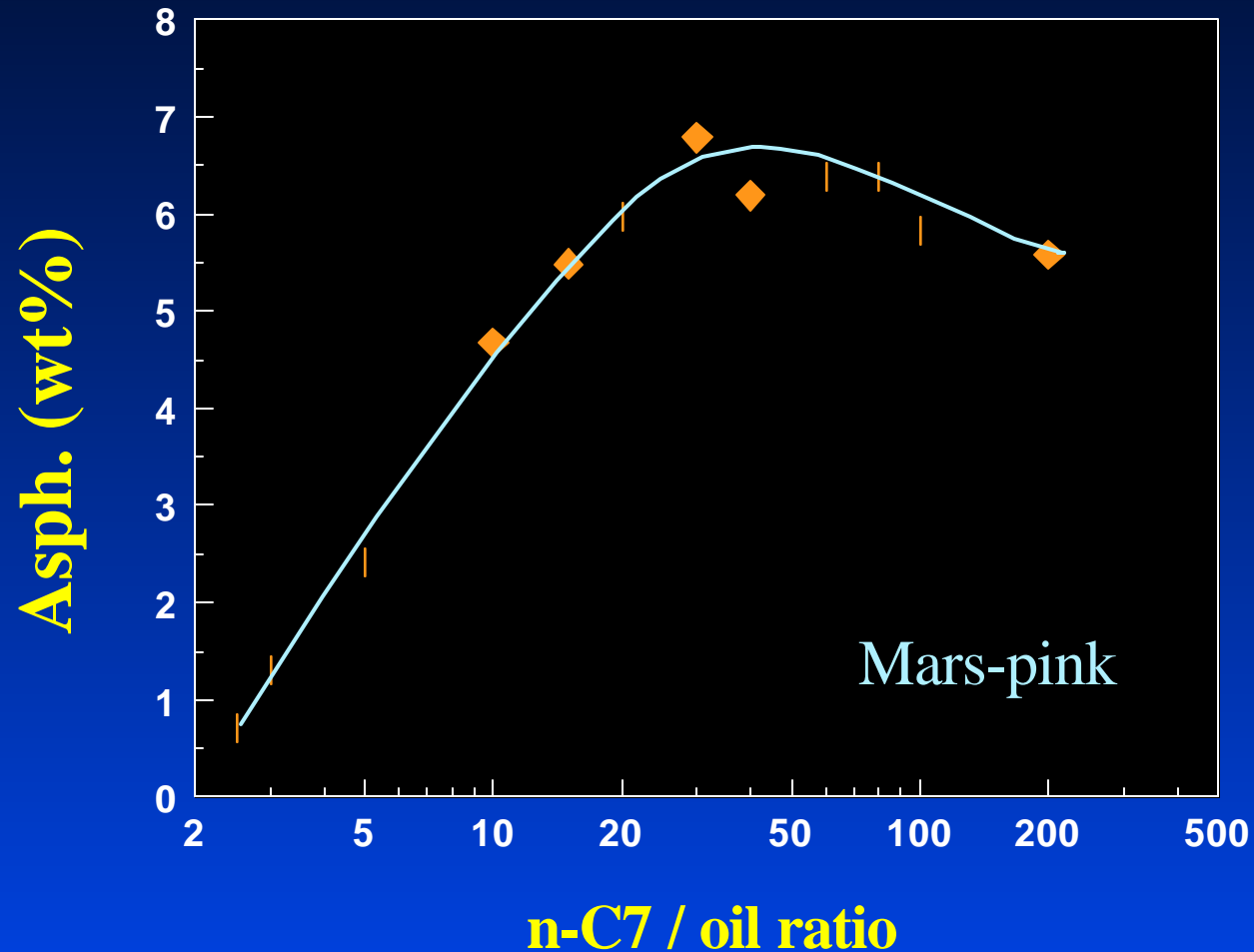
ASTM D2007-80:

- ▶ **Mixing crude oil / n-C7 (or n-C5) = 1 / 40**
- ▶ **Aging for two days**
- ▶ **Filtering through 0.22 mm filter paper**

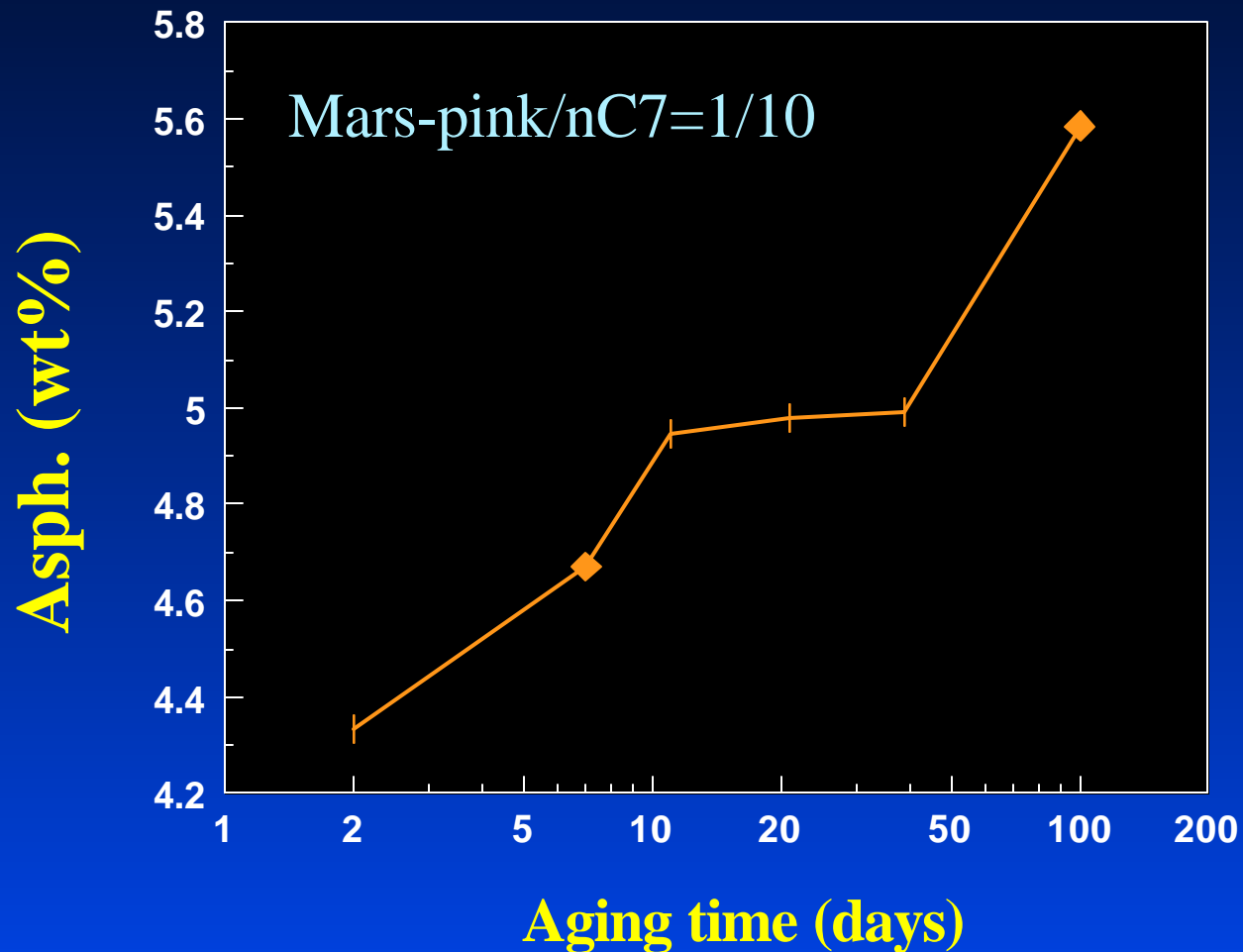
Separated asphaltene amount is determined by precipitant molecule size



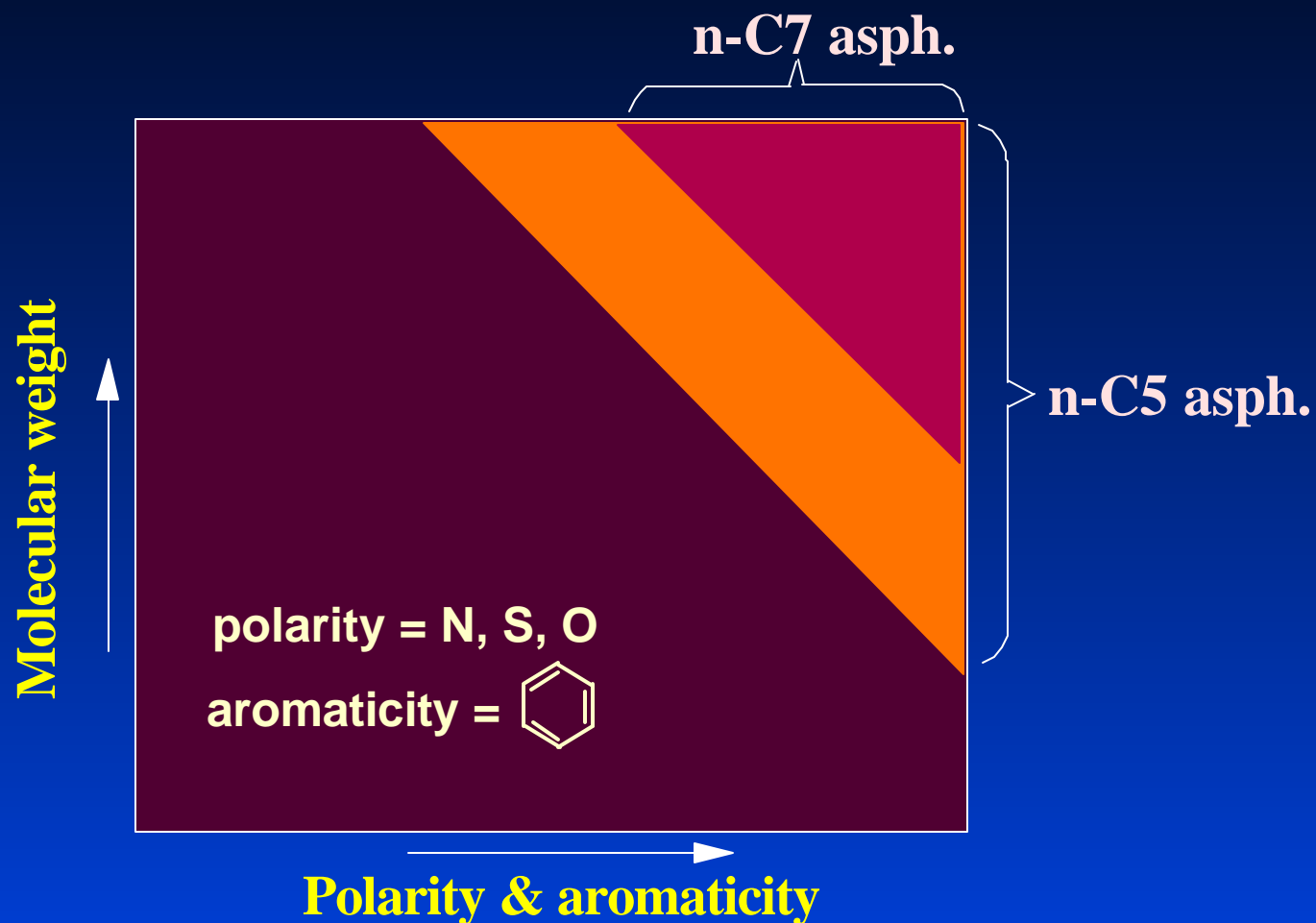
Separated asphaltene amount is also determined by precipitant/oil ratio



Furthermore, separated asphaltene amount is also time-dependent



Asphaltene is a compositional continuum



Long, R.B.: "The Concept of Asphaltenes," Chemistry of Asphaltenes, J.W. Bunger and N.C. Li (eds.), ACS, Washington, DC (1981) 17-27.

Typical asphaltene molecular properties

H/C = 0.8 - 1.4

Molecular weight:

depends on solvent and concentration

monomer = 500 - 1000

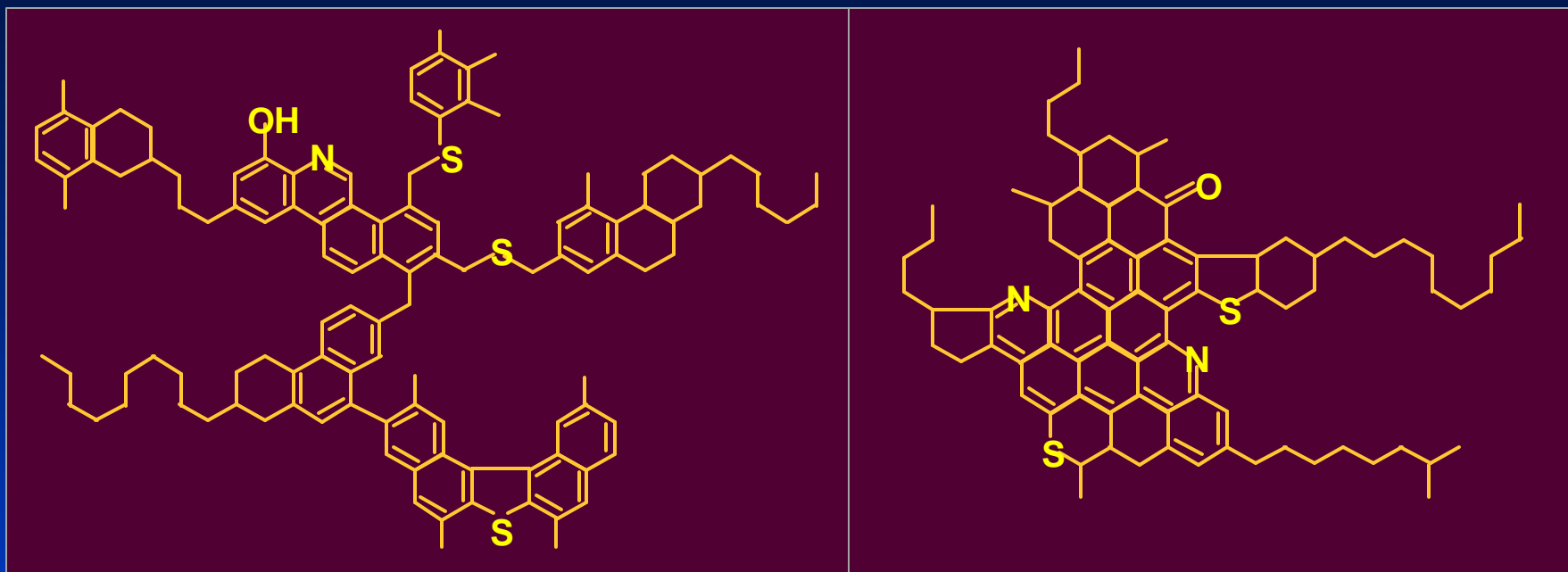
micelles = 1000 - 5000

Heteroatoms: acting as polar functional group

S=0.5-10 wt%; N=0.6-2.6 wt%; O=0.3-4.8 wt%

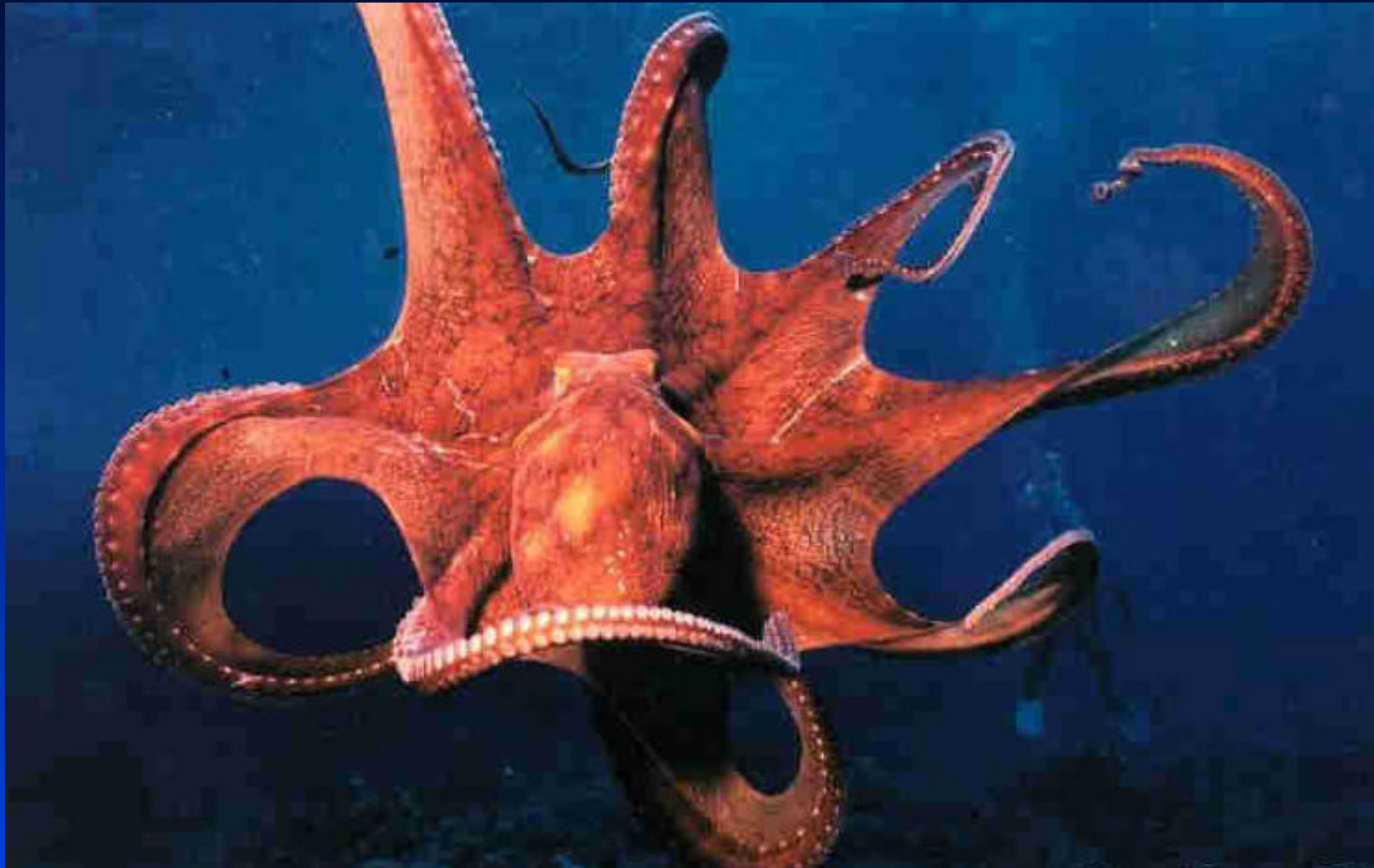
Metal elements: Ni, V, Fe

Hypothetical polycyclic structures for asphaltene molecules



Speight, J.G.: "A Chemical and Physical Explanation of Incompatibility during Refining Operations," Proc. 4th Intl. Conf. on the Stability and Handling of Liquid Fuels, US Dept. Energy, 169 (1992).

Asphaltene? Like a sea monster



Differences between asphaltene and wax

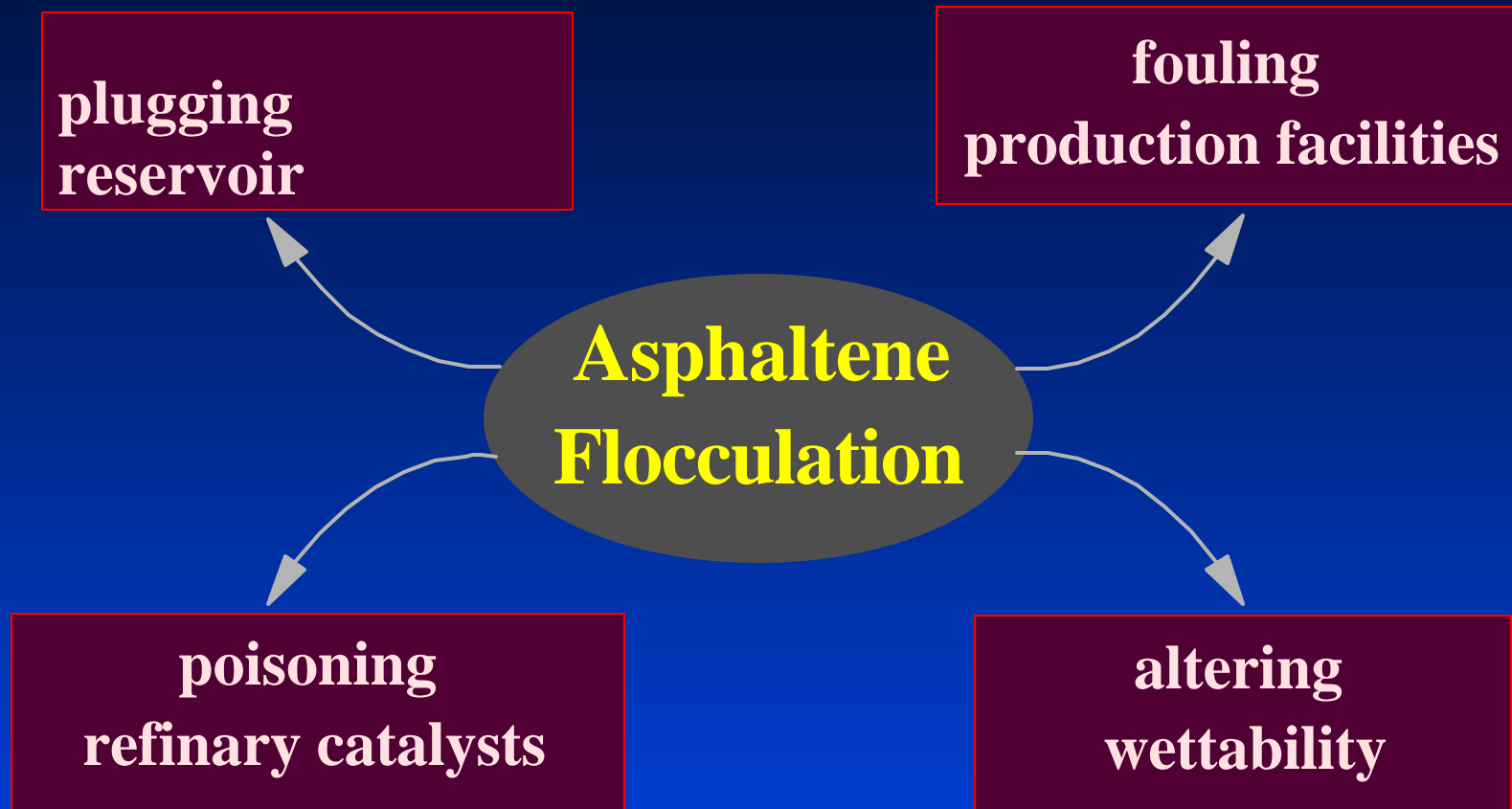
	Asphaltene	Wax
Dissolves in heptane:	No	Yes
Crystalline:	No	Yes
Melting point:	No	Yes

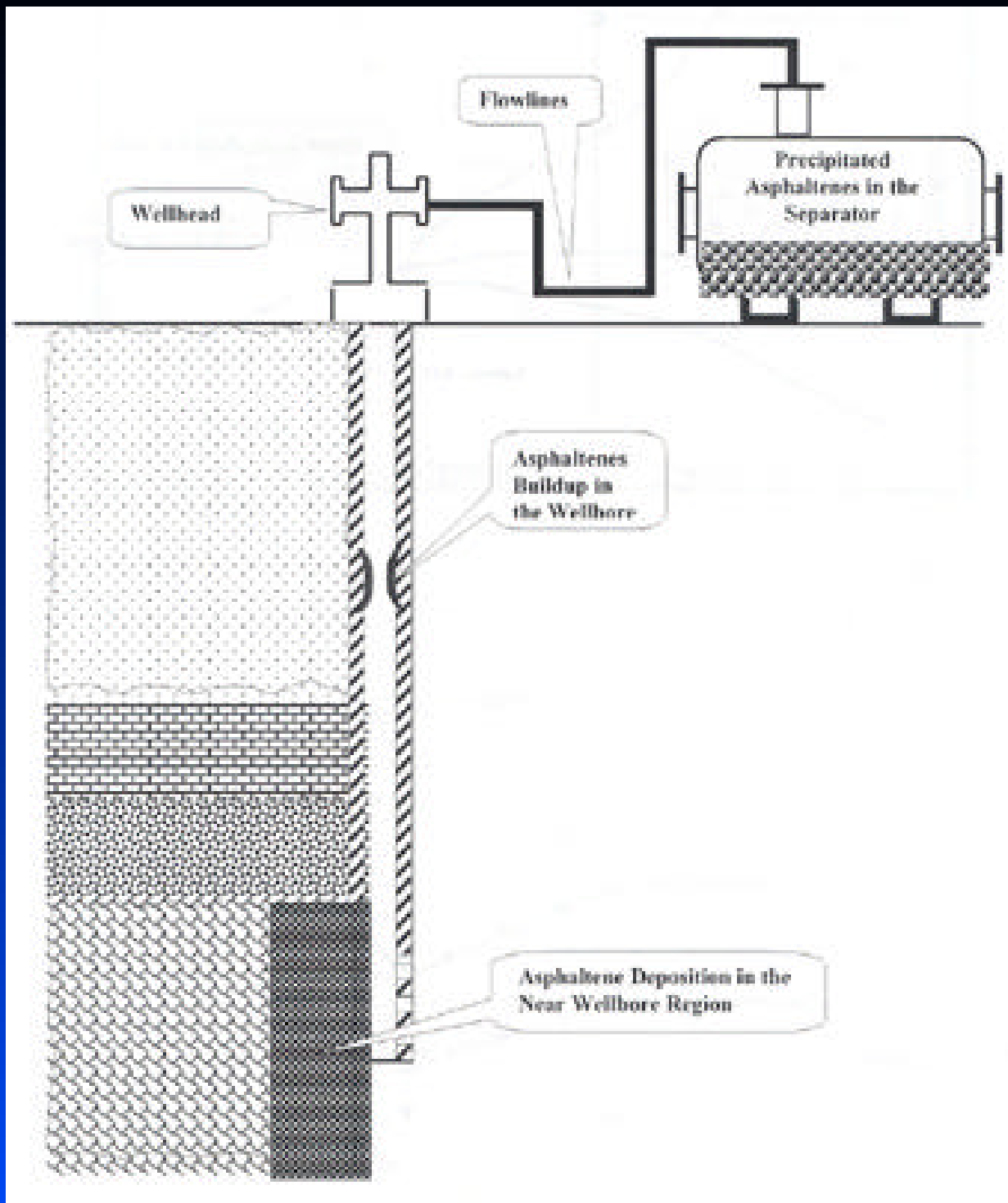
Buckley, J.S.: "Asphaltenes: Separating Fact from Fiction," SPE Distinguished Lecture Slides (1998-99).

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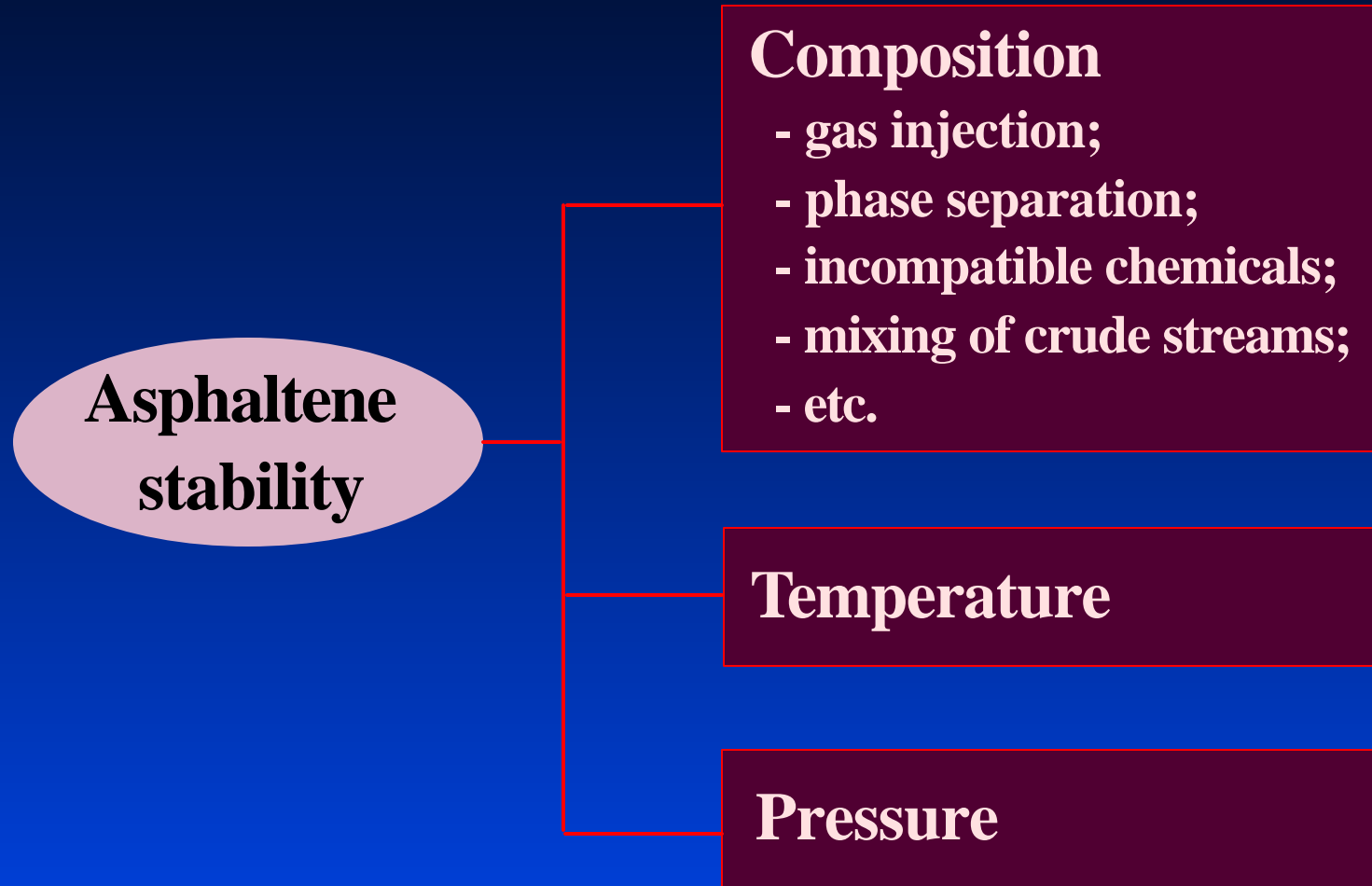
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What if asphaltene flocculates ?

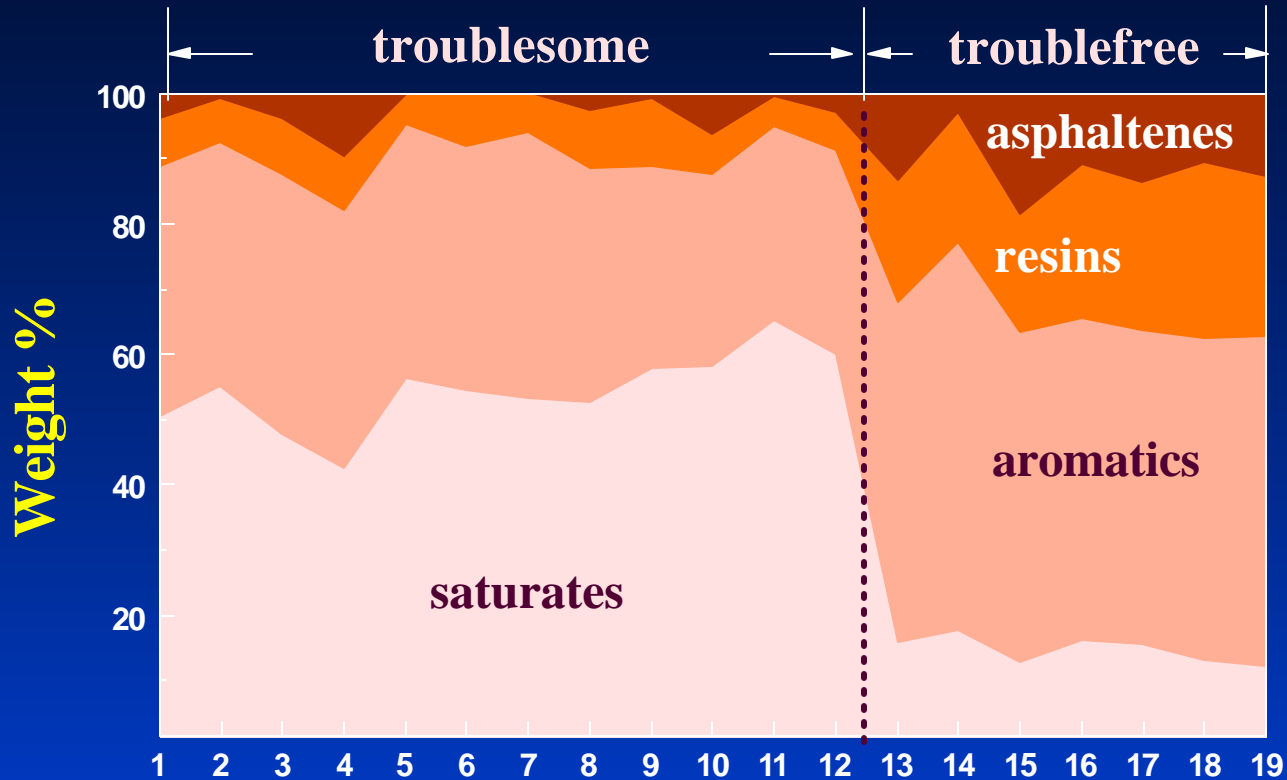




Stability is determined by crude oil solvency



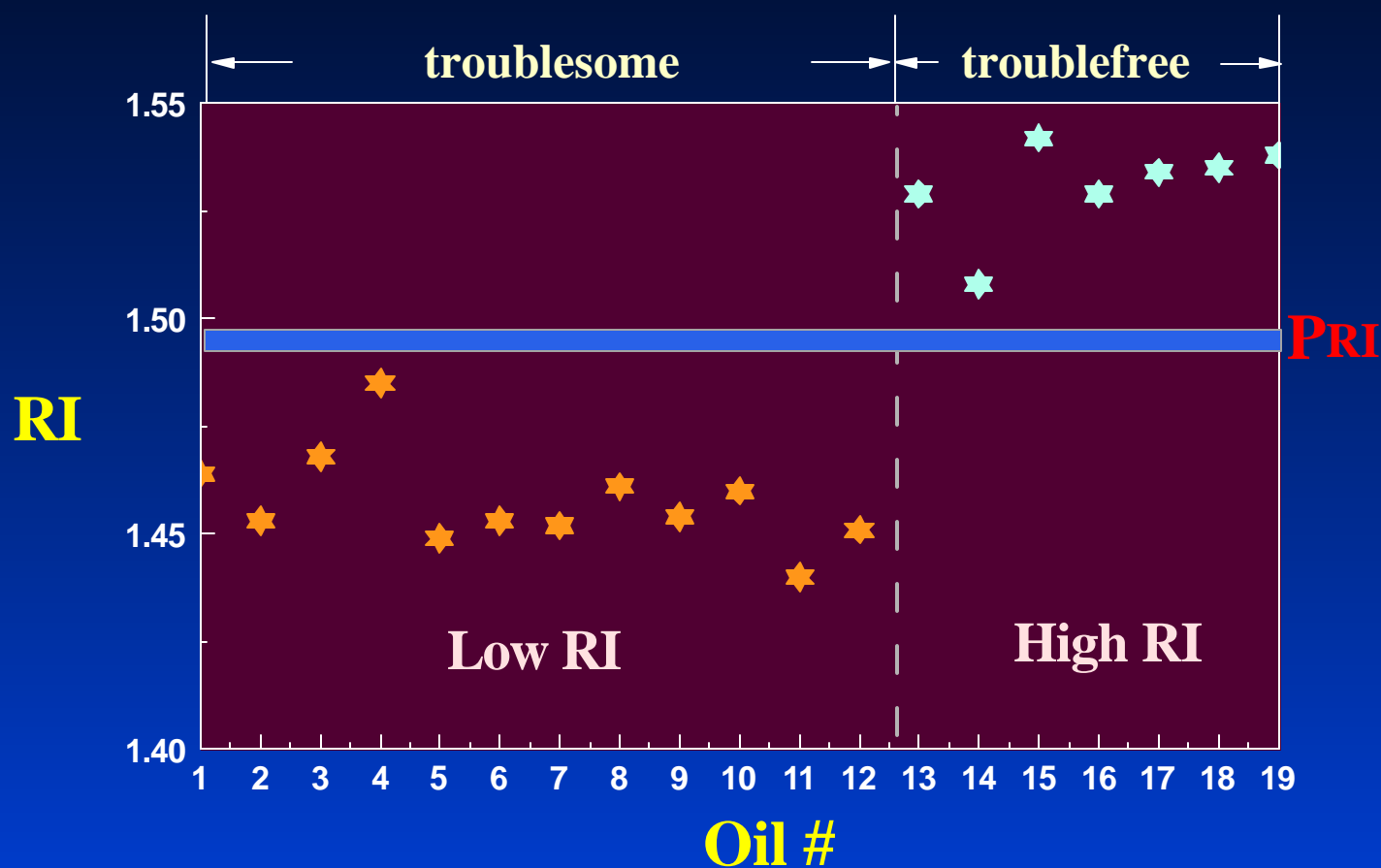
High asphaltene content is not necessary to associate with high risk of trouble



Oil # from Venezuela oil fields

Carbognani, L. and Espidel, J.: "Characterization of Solid Deposits from Production Facilities. Identification of Possible Causes of Deposits Formation," *Vision Technologica*, Vol. 3, No. 1, 35-42.

Asphaltene trouble is associated with RI

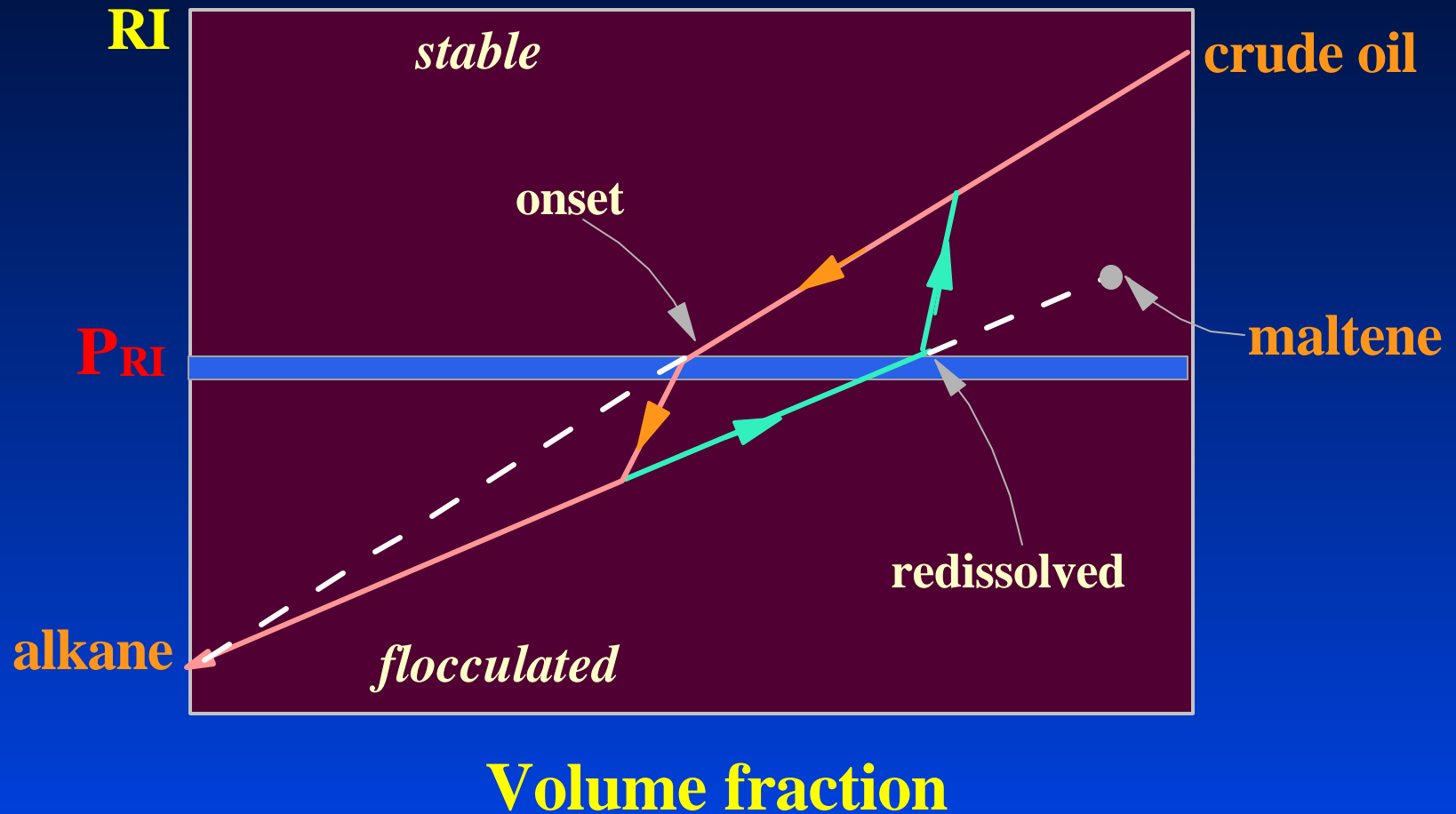


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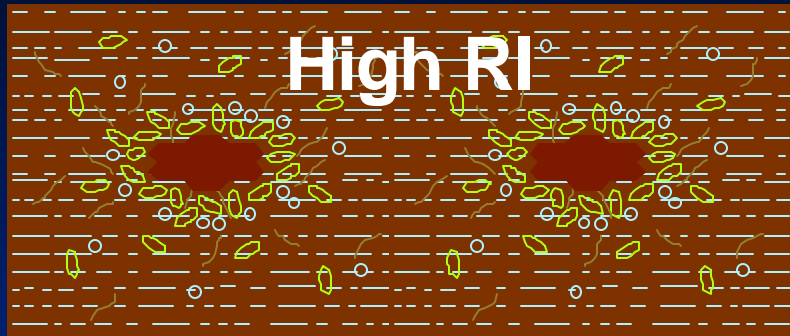
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For a given crude/precipitant, flocculation occurs at a critical RI



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Asphaltene stability varies with solvent RI



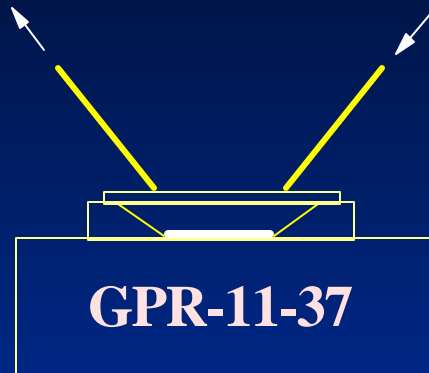
In a "good" solvent, asphaltenes are not strongly attracted to one another.

In a "poor" solvent, asphaltenes attractive forces are enhanced.



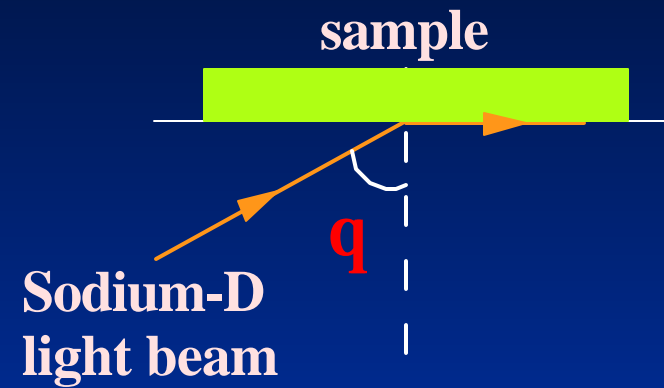
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RI measurement



refractometer

RI accuracy = 0.0001
T: 10 ~ 70 °C



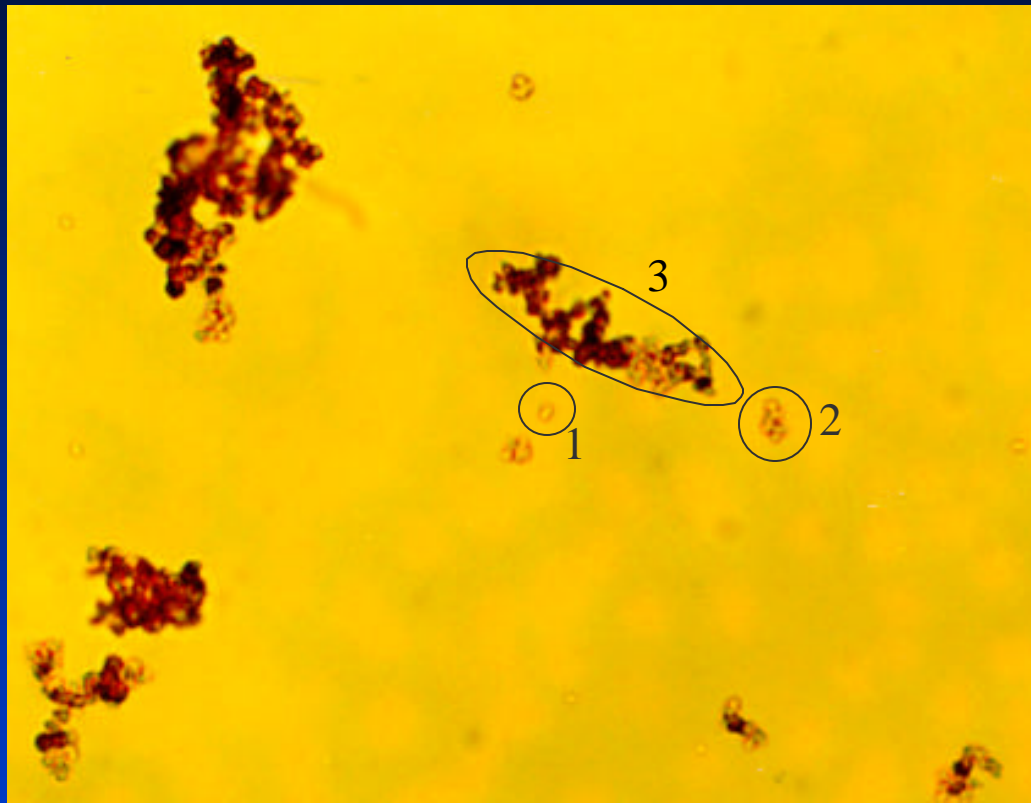
total internal reflection

$$n = \frac{1}{\sin(q)}$$

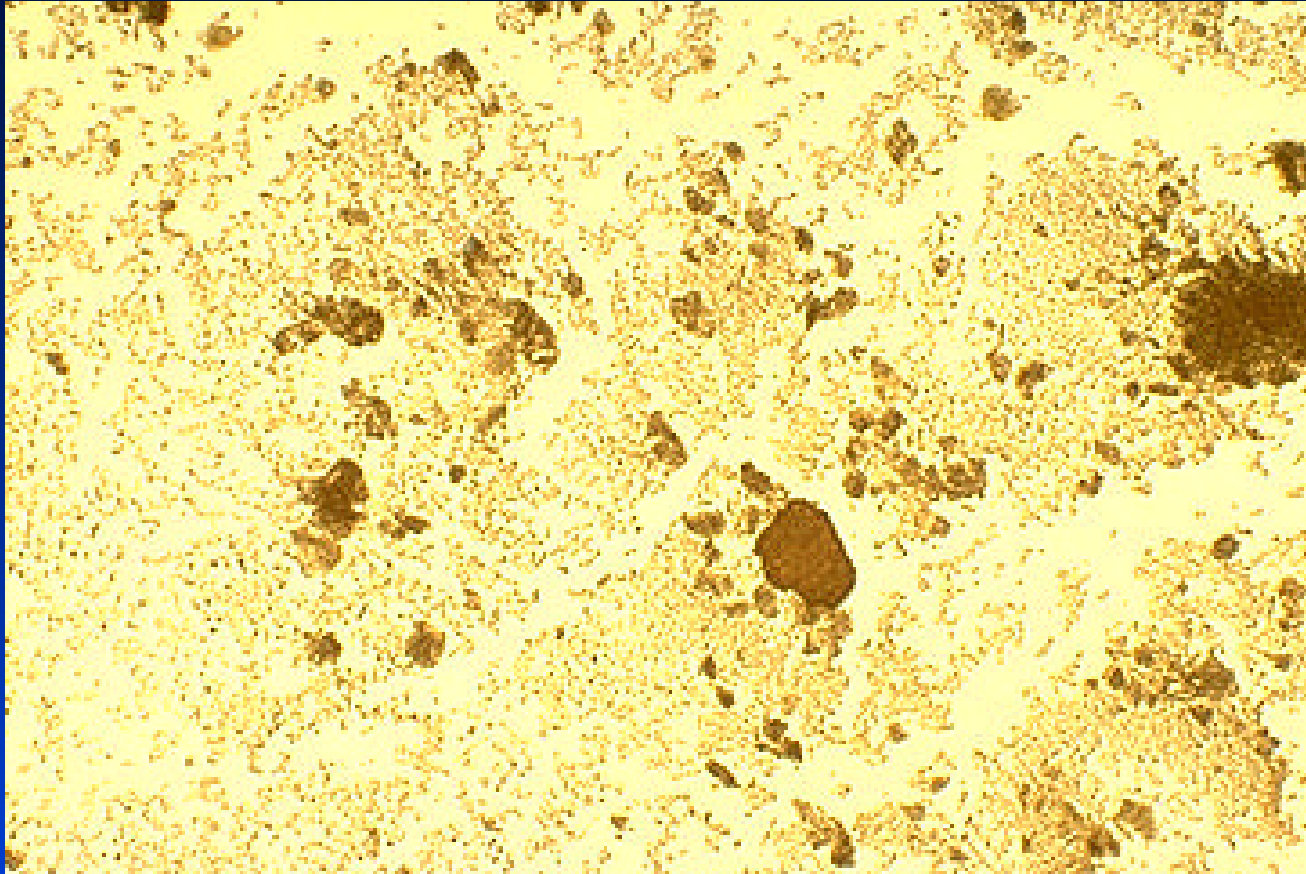
Procedure for onset detection

- ▶ **Prepare a series of mixtures with varying oil/precipitant ratios in sealed vials**
- ▶ **Aging for 24 hours**
- ▶ **Observe asphaltene precipitation, if any, under microscope**
- ▶ **Once onset is detected, record volume ratio and measure the refractive index (RI) for mixture**

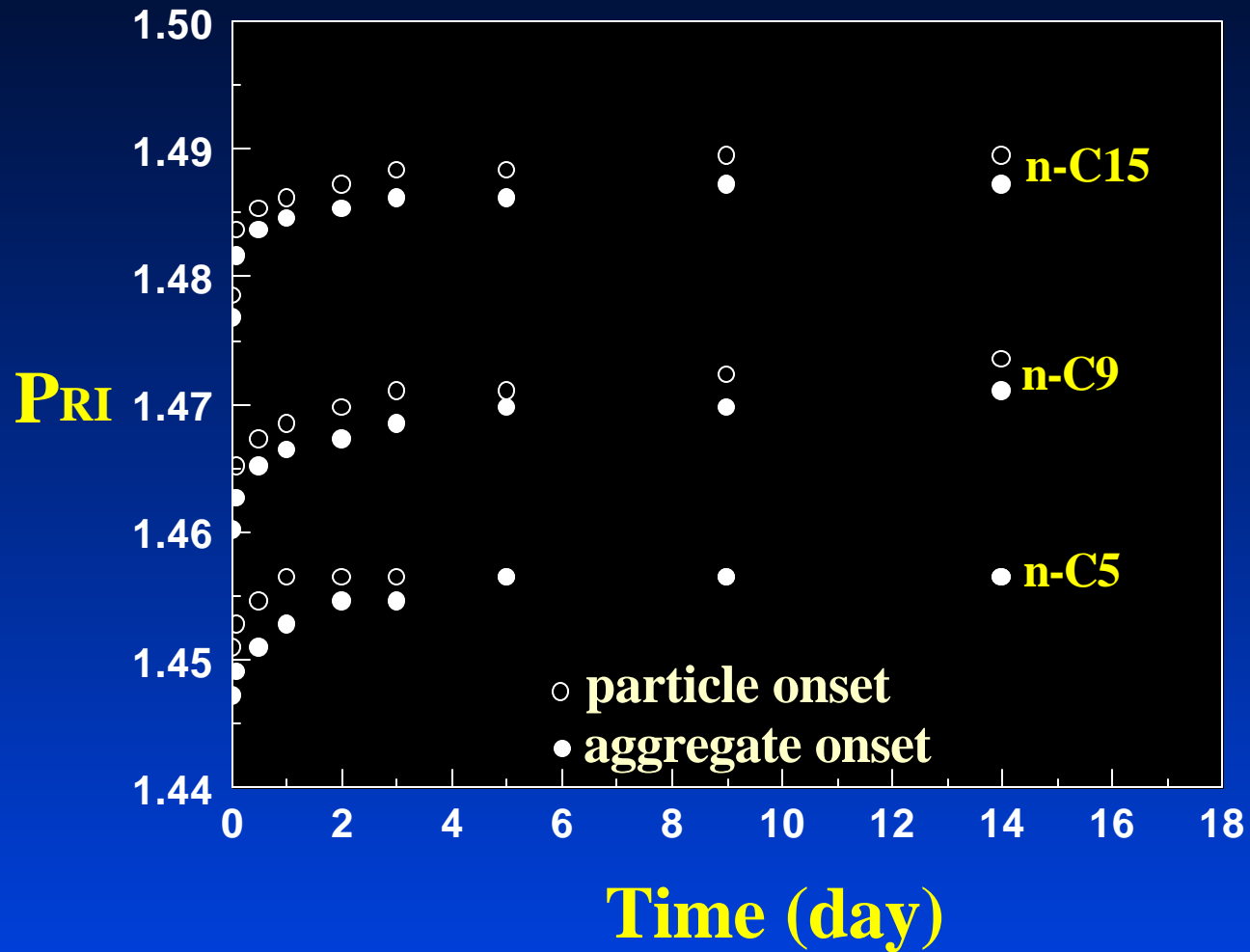
Asphaltene flocculates under microscope (1)



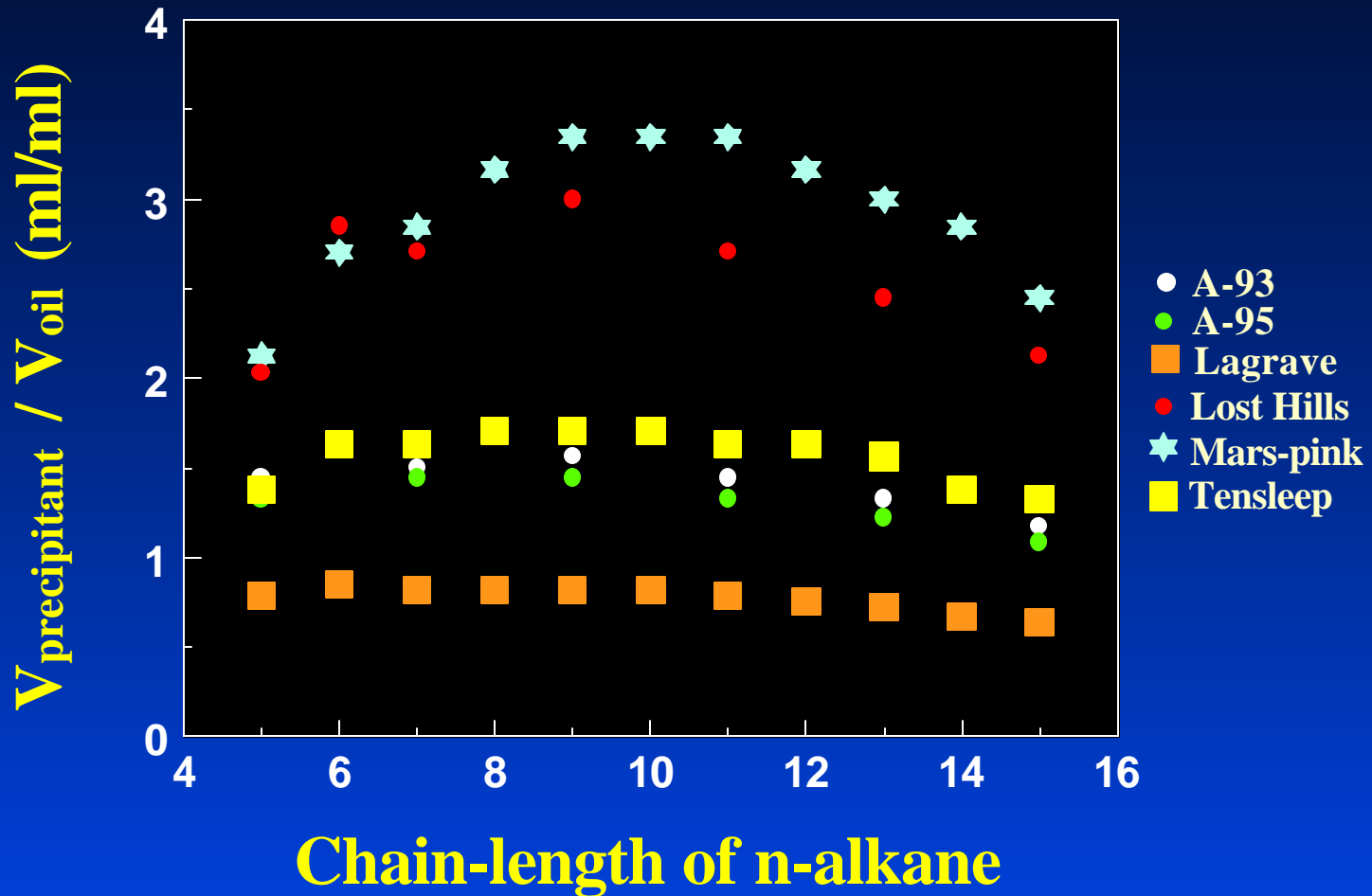
Asphaltene flocculates under microscope (2)



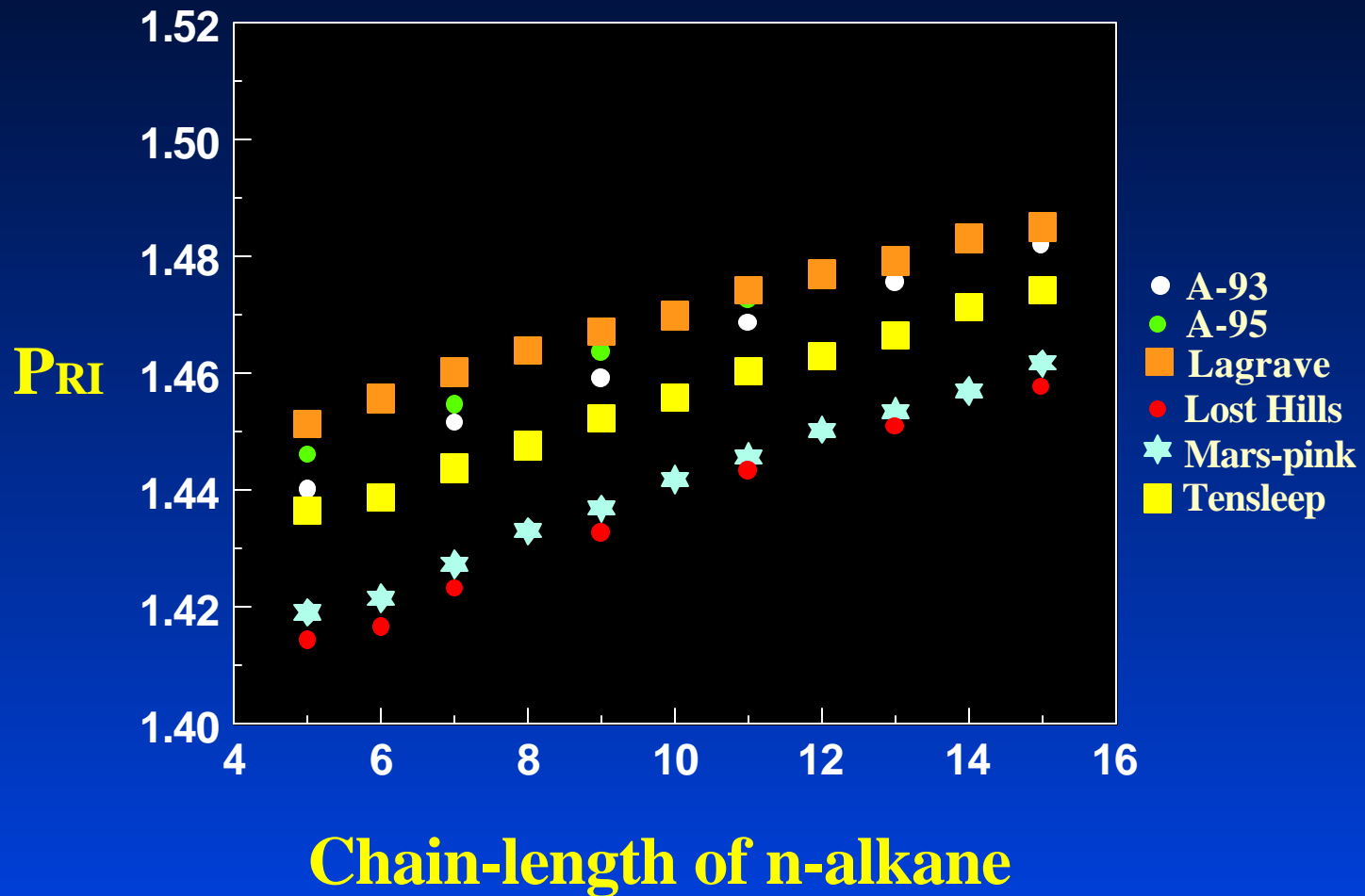
P_{RI} vs. time (Lagrange)



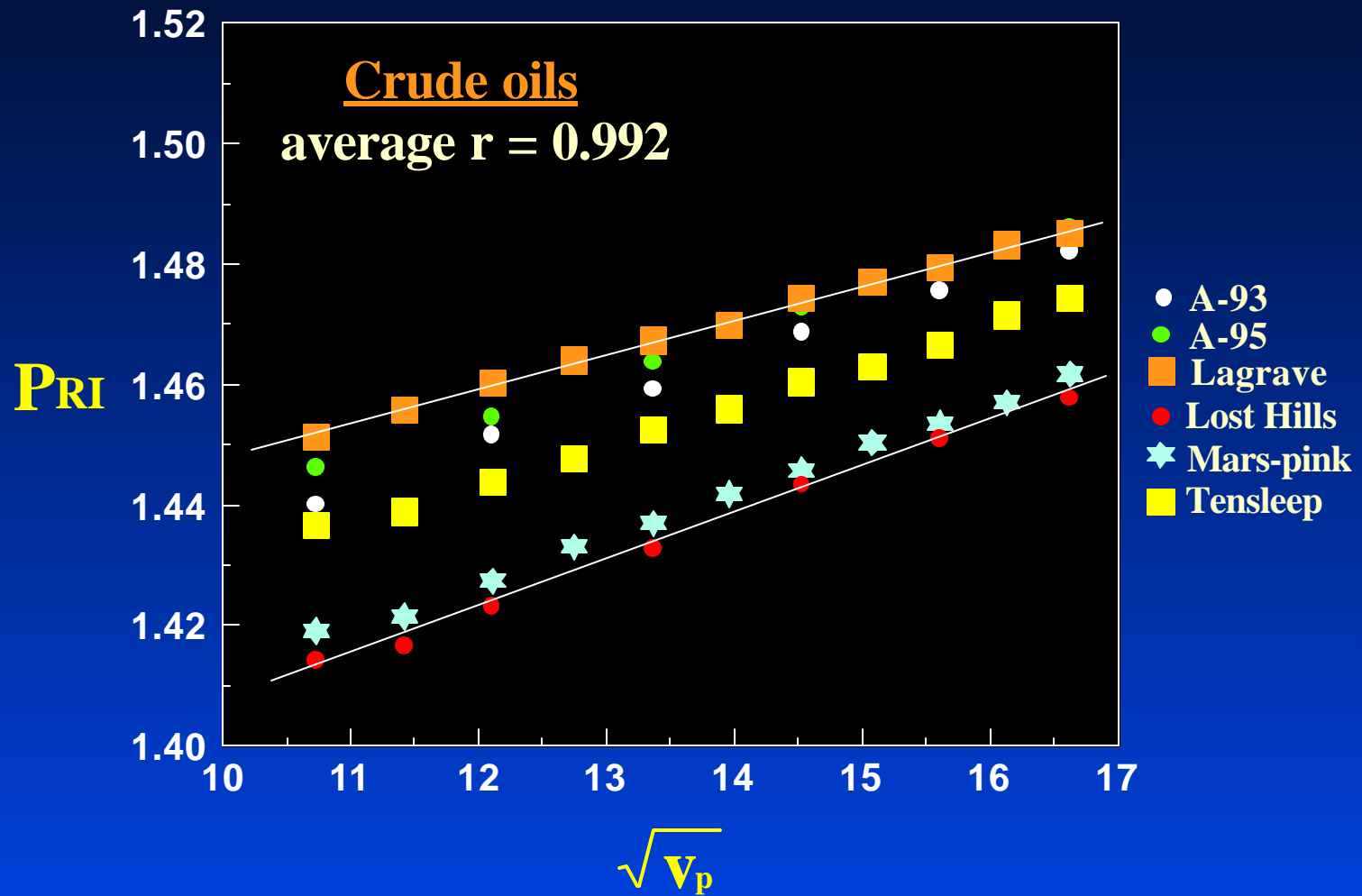
Volume ratios of precipitant to oil at onset



RI at onset (P_{RI}) for crude oils



Simple correlation of P_{RI} vs. precipitant



Conclusions

- ▶ **Asphaltene molecules are polyaromatics with side chains. They exist in crude oils as monomer and micelles equilibrated to each other. The sizes of asphaltene micelles fall in colloidal range.**
- ▶ **Asphaltene flocculation occurs when oil solvency is reduced. Flocculated asphaltene could have detrimental effects on oil production, transportation and refinery process.**
- ▶ **Refractive index (RI) is a simple, readily measurable parameter for characterizing crude oil solvency with respect to asphaltene stability.**